From atoms and molecules to new materials and technologies

Reduction of detailed kinetic mechanisms: Introduction to Mechanism Reduction Module for Chemical Workbench and best-practice

July 22, 2014



Overview of the webinar

- 1. About Kintech Lab
- 2. What is mechanism reduction and how it works?
- 3. Demo: working with Mechanism Reduction Module
- 4. Best practice recommendations for mechanism reduction
- 5. Demo: Setting reduction method parameters
- 6. Demo: Reducing mechanisms for a set of conditions
- 7. Demo: Application of several reduction methods



About Kintech Lab

KINTECH was founded in 2001 by scientists and engineers from and the Russian Research Center "Kurchatov Institute" and Moscow State University

Activity fields:

 Conducting of inventive research and consulting for a wide range of applications

- ✓ Software development for chemistry-intensive modeling and design in complete cycle
- Customer support in their own research activity using the advanced simulation capabilities of KINTECH's software



About Kintech Lab

Kintech Lab develops methods and special software tools for multilevel modeling in different engineering fields:

 Chemical Workbench – an integrated environment for the development and reduction of chemical mechanisms

 Khimera – a unique tool for calculating microscopic parameters from firstprinciples calculations

- KintechDB a network-based database for accumulation of Lab data on substances and processes
- \checkmark FDTD-II a tool for modeling the optical properties of metamaterials
- ✓ MD-kMC an integrated environment for atomistic modeling
 - EtchLab a tool for modeling and optimization of MEMS fabrication





The size of the detailed combustion kinetics mechanisms for surrogates of real fuels makes impossible massive design studies with CFD tools



Type of the mechanism	Applications
Detailed	 Hundreds of spx, thousands of rxns for typical hydrocarbon fuels Supposed to work in a wide range of conditions But should be tested
Skeletal	 Tens of spx, tens/hundreds of rxns for typical hydrocarbon fuels Derived from detailed by removing redundant species and reactions Derived for narrow range of conditions, sometimes for specific problems Should be tested at desired range of conditions
Reduced	 Tens of spx, tens of rxns for typical hydrocarbon fuels Derived from detailed/skeletal by application of QSS or QE approximations Derived for narrow range of conditions, sometimes for specific problems Should be tested at desired range of conditions
Global	 Can not be directly derived from previous one, just a very simple model Can be applied in a very restricted rage of conditions, only for one type of the problem MUST be tested every time before applications





Combustion process	Problem	Targets
Self-ignition	Shock tube/RCMFlow reactors	Ignition delay timeProfiles of species, temperature, pressure
Laminar Flame	Bunsen burnerFlat flame burnerDiffusion flame	 Laminar flame speed Extinction strain rate Profiles of species, temperature, pressure
Turbulent flame	Jet stirred reactor	Concentrations of species

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ABORATORY

Approach	Target for removal	Theoretical framework	Methods of reduction
Element or Species flux analysis	Species, reactions	Reaction paths	ROP, DRG, DRGEP
Time-scale analysis	Species, reactions	CSP	Time scale analysis, importance index analysis
Sensitivity analysis	Species, reactions	PCA	Species or rates sensitivity matrix analysis



Demo: working with Mechanism Reduction Module

🕞 Me	📲 Mechanism Reduction: 'JetA high temperature detailed mechanism'						
	Pro	cesses	Includ	le By Process	•	Models	
275	0+0+M<=>0	$_2 + M$	V	☑ 1		DRG	
276	0+H+M<=>0	H+M	V	☑ 1			Run
277	OH + OH + M<=	=>H ₂ O ₂ + M		☑ 1		Property	Value
278	$H + O_2 + M \leq > H$	$HO_2 + M$	V	V 1		Substances	O2 H2O H2 C10H22/;;;n/
279	$H + O_2 + N_2 <=>H$	$HO_2 + N_2$		Reaction pathway diagram		Lindex/max_index_ratio_thr	
 Library of state-of-art reduction algorithms (DRG, CSP, ROP, PCA) Integration with Chemical Workbench 							
22	C ₅ H ₁₀ (1)		☑ 1		M		 ▷ CH3OH Σ = ▷ C3H5(a) Σ =
23	C ₅ H ₁₁ (1)		☑ 1	Сн.сно нсо с	Cife Ci	C.H _s (1)	 ▷ ♥ CH3O Σ = ▷ ♥ C3H4(a) Σ = ▷ ♥ C5H11(1) Σ =
24	C ₅ H ₄ O	V	0.951338				
25	C_5H_5	V	0.999977	HC CH ₂ CO H	СН,С	С.Н.	▷ ☑ C2H6 ∑ = ▷ ☑ C2H5 ∑ =
26	C_5H_6		0.061886				$ \overrightarrow{\nabla} C2H3 \qquad \Sigma = $
27	C ₅ H ₉		☑ 1	C: CH:(s) CO C:H] [C.H.	 ▷ ♥ C6H13(1) Σ = ▷ ♥ CH2CHO Σ = ▷ ♥ C7H15(1) Σ =
28	C6H11		0.738139				 ▷ ♥ CH3CH0 Σ = ▷ ♥ C7H14(1) Σ = ▷ ♥ C7H12 Σ =
29	C ₆ H ₁₂ (1)		☑ 1			O	V V C/H13 2 []=
30	C ₆ H ₁₂ (2)		☑ 1			500 4.96227e-05 200	par: Temperature = 1.400000E+03
31	C ₆ H ₁₂ (3)	V	0.960339		Ŧ		· · · · · · · · · · · · · · · · · · ·
Reaction Path Save X Close							



Test problem:

- Auto-ignition
- stoichiometric n-heptane/air
- 1 bar
- 650 2000K

Kinetic mechanism:

 Curran, H. J., P. Gaffuri, W. J. Pitz, and C. K. Westbrook, "A Comprehensive Modeling Study of n-Heptane Oxidation" Combustion and Flame 114:149-177 (1998)

Target:

- Ignition delay time
- 30% maximum simulation error in comparison with detailed kinetic mechanism



Setting numerical parameters for reduction method



Threshold value	Iteration #1(err)	Iteration #2(err)	Iteration #3(err)
0.002	-207(err 0.02%)	-21(err 0.02%)	-2(err 0.02%)
0.004	-241(err 0.02%)	-28(err 0.02%)	-4(err 0.02%)
0.008	-266(err 0.02%)	-27(err 0.02%)	-4(err 0.02%)
0.016	-301(err 0.02%)	-25(err 0.02%)	-2(err 0.02%)
0.024	-327(err 0.03%)	-26(err 0.03%)	-1(err 0.03%)
0.036	-392(err 0.2%)	-36(err 0.3%)	-4(err 0.3%)
0.054	-410(err 1.7%)	-38(err 1.7%)	-1(err 1.7%)
0.081	-413(err 1.5%)	-33(err 0.7%)	-4(err 0.7%)
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Degree	e of reduction'	Error of IDT simulation	

Setting numerical parameters for reduction method





Reducing mechanisms for a range of conditions

Low-to-high temperature ignition and NTC: How much points should we select for mechanism reduction, keeping the accuracy of reduced mechanism acceptable?





Reducing mechanisms for a range of conditions



For efficient generation of the reduced kinetic mechanism in a range of conditions it is enough to take a single point in every characteristic interval of initial conditions. For example, one point in low-temperature region and one point in high-temperature region



Application of several reduction methods

STEP 1

Reduction of redundant species:

- Only species, important at given conditions, will be present in mechanism
- Also a lot of reactions will be removed with species

Connectivity based methods: DRG, DRGEP



STEP 2

Reduction of redundant reactions

Still redundant reactions can be removed from mechanism with important species

Time-scale analysis methods: CSP-based Sensitivity analysis methods: PCA-, PCAF-based









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